

A NEW EQUATION OF STATE FOR SOLID *para*-HYDROGEN

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Solid *para*-H₂ is a popular accommodating host for impurity spectroscopy due to its unique softness and the spherical symmetry of *para*-H₂ in its $J=0$ rotational level.^{a,b} To simulate the properties of impurity-doped solid *para*-H₂, a reliable model for the ‘soft’ pure solid *para*-H₂ at different pressures is highly desirable. While a couple of experimental^c and theoretical^d studies aimed at elucidating the equation of state (EOS) of solid *para*-H₂ have been reported, the calculated EOS was shown to be heavily dependent on the potential energy surface (PES) between two *para*-H₂ that was used in the simulations.^e The current study also demonstrates that different choices of the parameters governing the Quantum Monte Carlo simulation could produce different EOS curves.

To obtain a reliable model for pure solid *para*-H₂, we used a new 1-D *para*-H₂ PES reported by Faruk *et al.*^f that was obtained by averaging over Hinde’s highly accurate 6-D H₂–H₂ PES.^g The EOS of pure solid *para*-H₂ was calculated using the PIMC algorithm with periodic boundary conditions (PBC). To precisely determine the equilibrium density of solid *para*-H₂, both the value of the PIMC time step τ and the number of particles in the PBC cell were extrapolated to convergence. The resulting EOS agreed well with experimental observations, and the *hcp* structured solid *para*-H₂ was found to be more stable than the *fcc* one at 4.2K, in agreement with experiment. The vibrational frequency shift of *para*-H₂ as a function of the density of the pure solid was also calculated, and the value of the shift at the equilibrium density is found to agree well with experiment.

^a T. Momose, H. Honshina, M. Fushitani and H. Katsuki, *Vib. Spectrosc.* **34**, 95(2004).

^b M. E. Fajardo, *J. Phys. Chem. A* **117**, 13504 (2013).

^c I. F. Silvera, *Rev. Mod. Phys.* **52**, 393(1980).

^d F. Operetto and F. Pederiva, *Rhys. Rev. B* **73**, 184124(2006).

^e T. Omiyinka and M. Boninsegni, *Rhys. Rev. B* **88**, 024112(2013).

^f N. Faruk, M. Schmidt, H. Li, R. J. Le Roy, and P.-N. Roy, *J. Chem. Phys.* **141**, 014310(2014).

^g R. J. Hinde, *J. Chem. Phys.* **128**, 154308(2008).